

Ze: an Effective Atomic Number for Polychromatic X-ray Spectra

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- What is an “effective atomic number?”
- Cross-sections describe materials.
- Existing approaches to Z_{eff} are inadequate
- A new technique, {Ze, RhoE}, provides an improved approach to characterizing materials with an effective atomic number.
- An Example demonstrates the practical use of {Ze, RhoE}.
- An Application is available to Calculate {Ze, RhoE}.

- **An effective atomic number:**
 - Alias: Effective-Z or Z_{eff}
 - “Describes” a complex (non-elemental) material (compound, mixture) as a single number
 - Monoenergetic or **Polychromatic spectra**
 - Examples Include:
 - Radiography, Computed Tomography (CT)
 - Multi-spectral Radiography
 - **Multi-spectral or Dual-energy CT (DECT)**
 - Dose to tissue or other materials



Beers-Lambert Law for Polychromatic Spectra

$$I[x] = \int dEx S[Ex] \text{Exp}[-(\mu/\rho)[Z, Ex] \rho_m x]$$

(μ/ρ) is the mass-absorption coefficient (cm^2/g)

Z is the atomic number

Ex is the x-ray energy

ρ_m is the mass density

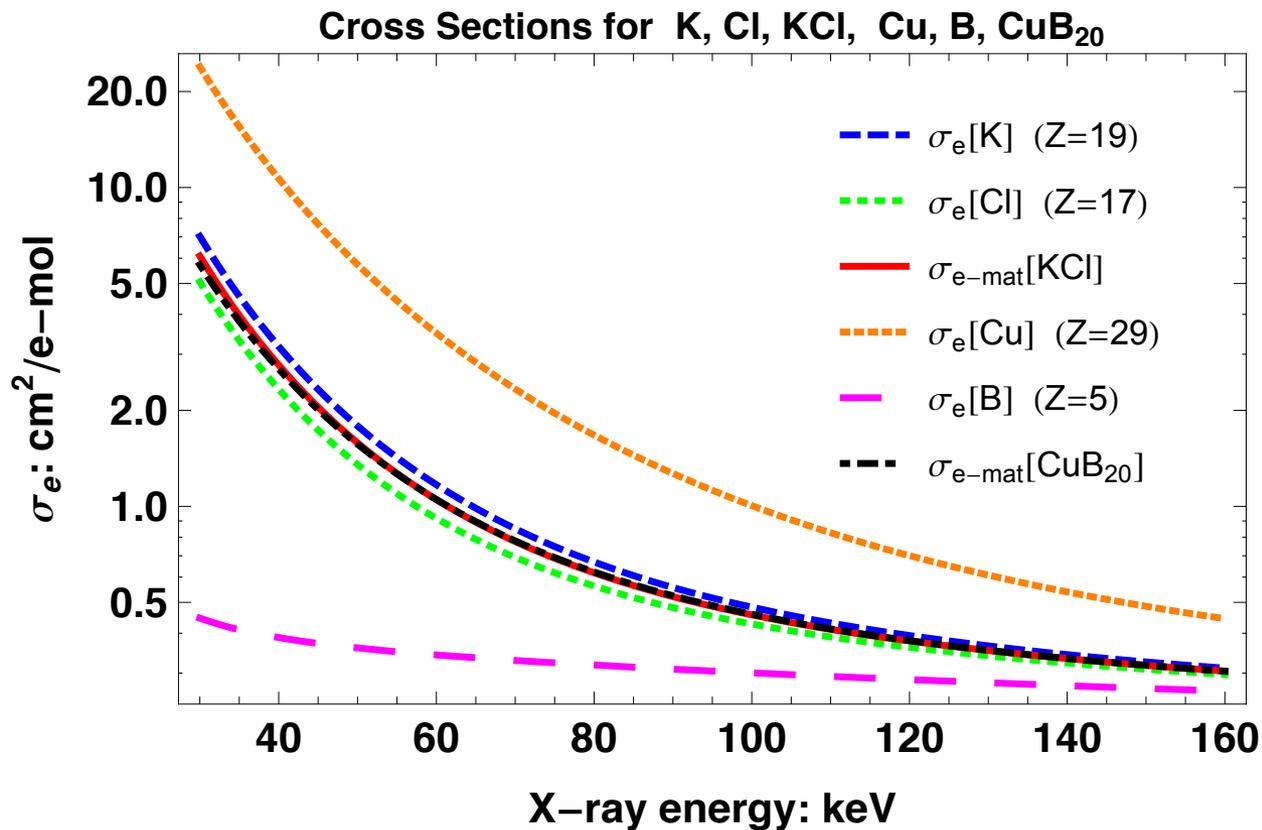
x is the length of the transmission path

$$I[x] = \int dEx S[Ex] \text{Exp}[-\sigma_e[Z, Ex] \rho_e x] \quad \text{Equation 1}$$

σ_e is the mass-absorption coefficient ($\text{cm}^2/\text{e-mol}$)

ρ_e is the electron density ($\text{e-mol}/\text{cm}^3$)

$I[X]$ is the energy-integrated transmission if: $\int dEx S[Ex] = 1.0$





For a compound with N elements:

$$\sigma_{e-mat}[Ex] = \sum_{i=1}^N \alpha_i \sigma_e[Z_i, Ex] ,$$
$$\alpha_i = n_i Z_i / \sum_{j=1}^N n_j Z_j .$$

Equation 2

σ_{e-mat} is the electronic cross section of the material

Z_i, n_i are the Z and number of that element in the compound

α_i is the fraction of each mole of electrons provided by element Z_i

We need an effective-Z description of the molecule

$$\sigma_e[Z_{eff}, Ex] \cong \sigma_{e-mat}[Ex]$$

Equation 3

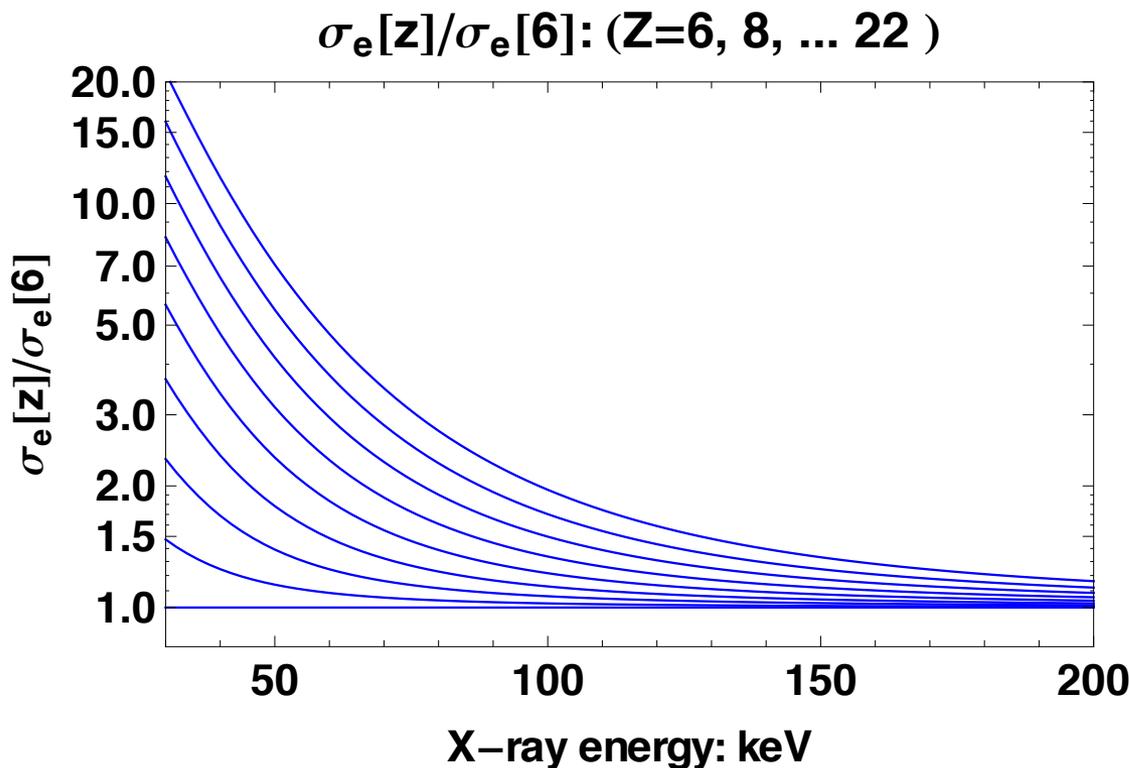
(This equation can only be exact for single-element materials)

- **Mayneord (1937)** (interpolation formula) [1]

$$Z_{eff} = \sqrt[p]{\sum_i^N \alpha_i Z_i^p}$$

Equation 4

- Typically $3 \leq p \leq 4$ (depends on Ex and Z_i)
 - Very common (ubiquitous)
 - Cannot write $\sigma_e[Z_{eff}, Ex]$
- **Several other versions** [2–5]
 - Provide: $Z_{eff}[Ex]$
 - Cannot write $\sigma_e[Z_{eff}, Ex]$



- At high keV, cross-section per e-mol becomes independent of Z
- Low-energy end of spectrum carries most of the Z-dependence
- Ze is simply interpolated into the surface of $\sigma_e[Z, Ex]$

$$\sigma_{Z_e}[Ex] \stackrel{\text{def}}{=} \sigma_e[Z_e, Ex] = (1 - \text{frac})\sigma_e[Z_1, Ex] + \text{frac} \sigma_e[Z_1 + 1, Ex]$$

$$Z_1 = \text{floor}[Z_e] \text{ and } \text{frac} = \text{fractional_part}[Z_e] \quad \text{Equation 5}$$

- $\sigma_{Z_e}[Ex]$ is interpolated into $\sigma_e[Z, Ex]$, using the fractional contribution of electrons from Z_1 and Z_2 as the interpolation variable.
- $\sigma_e[Z, Ex]$ (cm²/e-mol) is converted from tabulated cross sections [6–8] $\sigma[Z, Ex]$ (b/atom), or (μ/ρ) (cm²/g) and the atomic weight, A:

$$\sigma_e[Z, Ex] = \frac{K_0}{Z} \sigma_e[Z, Ex] = \frac{A}{Z} (\mu/\rho)[Z, Ex] \quad \text{Equation 6}$$

$$K_0 = 0.6022 \text{ cm}^2/\text{b} - \text{mol} .$$

- ρ_e and m_e can be calculated from the volumetric mass density, ρ_m , or the areal mass density, $m_m = \rho_m \times$

$$\rho_e = \rho_m \sum_{i=1}^N w_i Z_i / A_i$$

$$m_e = m_m \sum_{i=1}^N w_i Z_i / A_i , \quad \text{Equation 7}$$

$$w_i = n_i / \sum_{j=1}^N n_j .$$

- Assigning a Ze to a particular material is done with a least squares fit.
- Minimizing the overall error in the cross section overly emphasizes the low energies where the cross-section differences are greatest, but there is little x-ray transmission.
- It is better to find the Ze that generates the least mean-squared error in transmission:

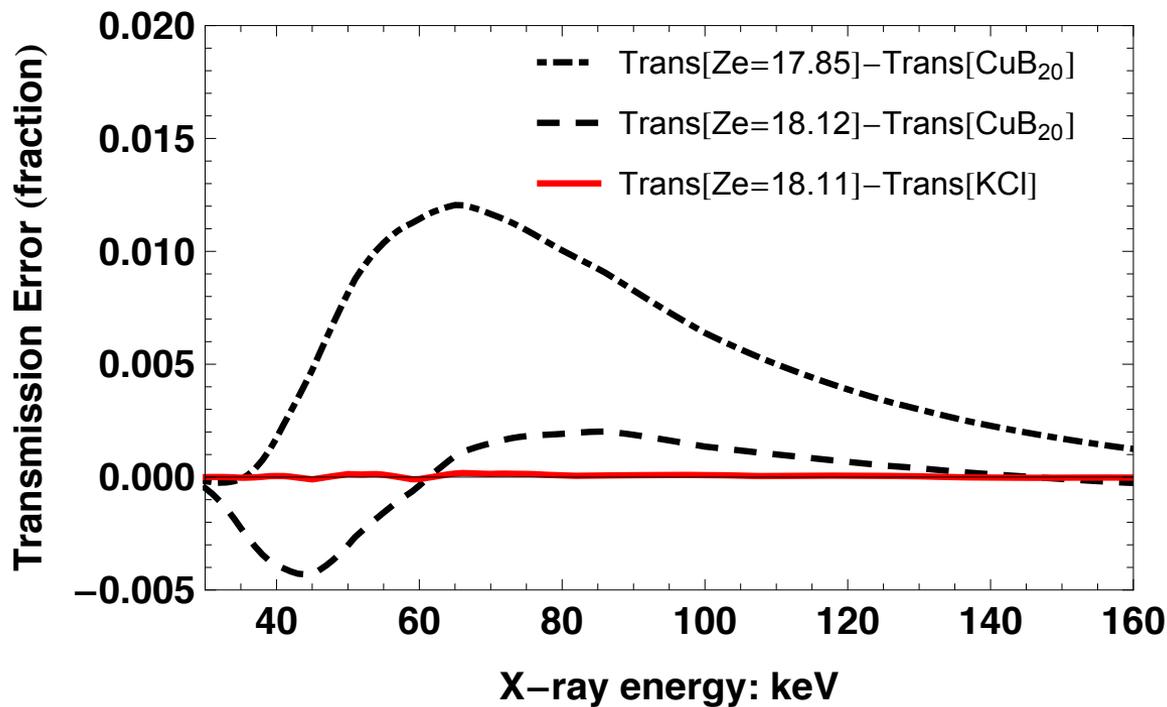
$$\sum_{eMin}^{eMax} (Trans_{e-mat}[Ex] - Trans_{Ze}[Ex])^2 \rightarrow \text{minimum}$$

Equation 8

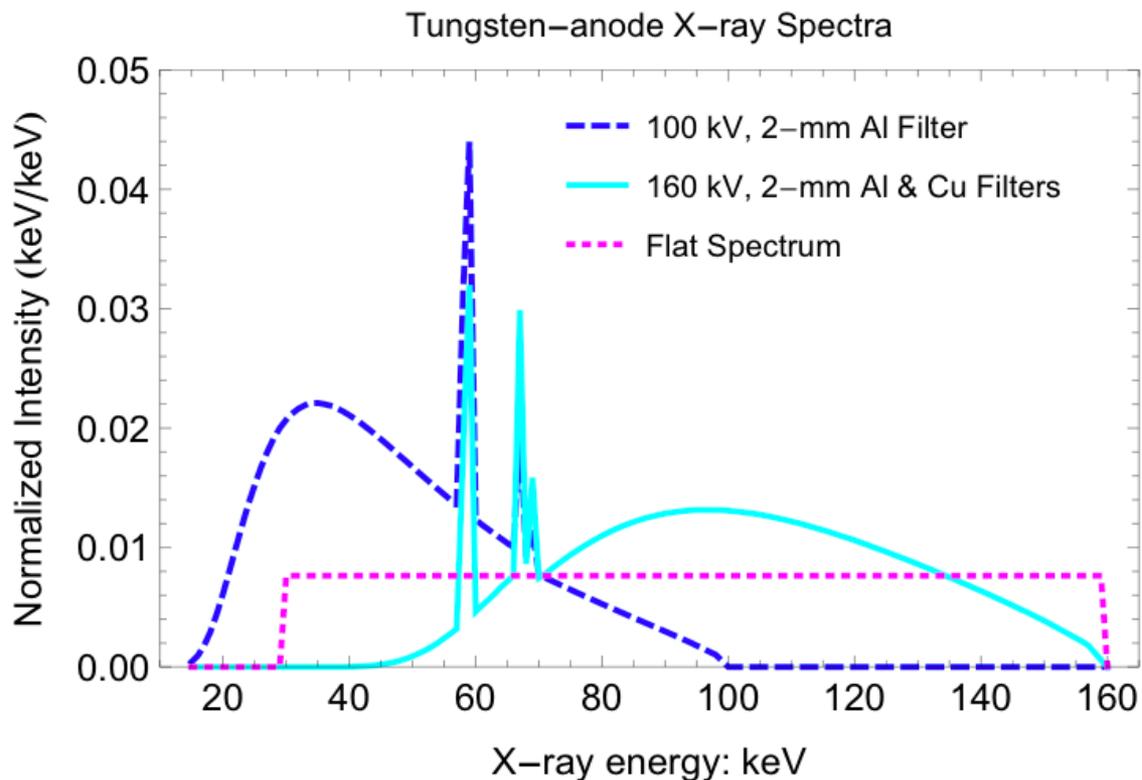
$$Trans_{e-mat}[Ex] = \text{Exp}[-\sigma_{e-mat}[Ex] m_{e-samp}]$$

$$Trans_{Ze}[Ex] = \text{Exp}[-\sigma_{Ze}[Ex] m_{e-samp}] .$$

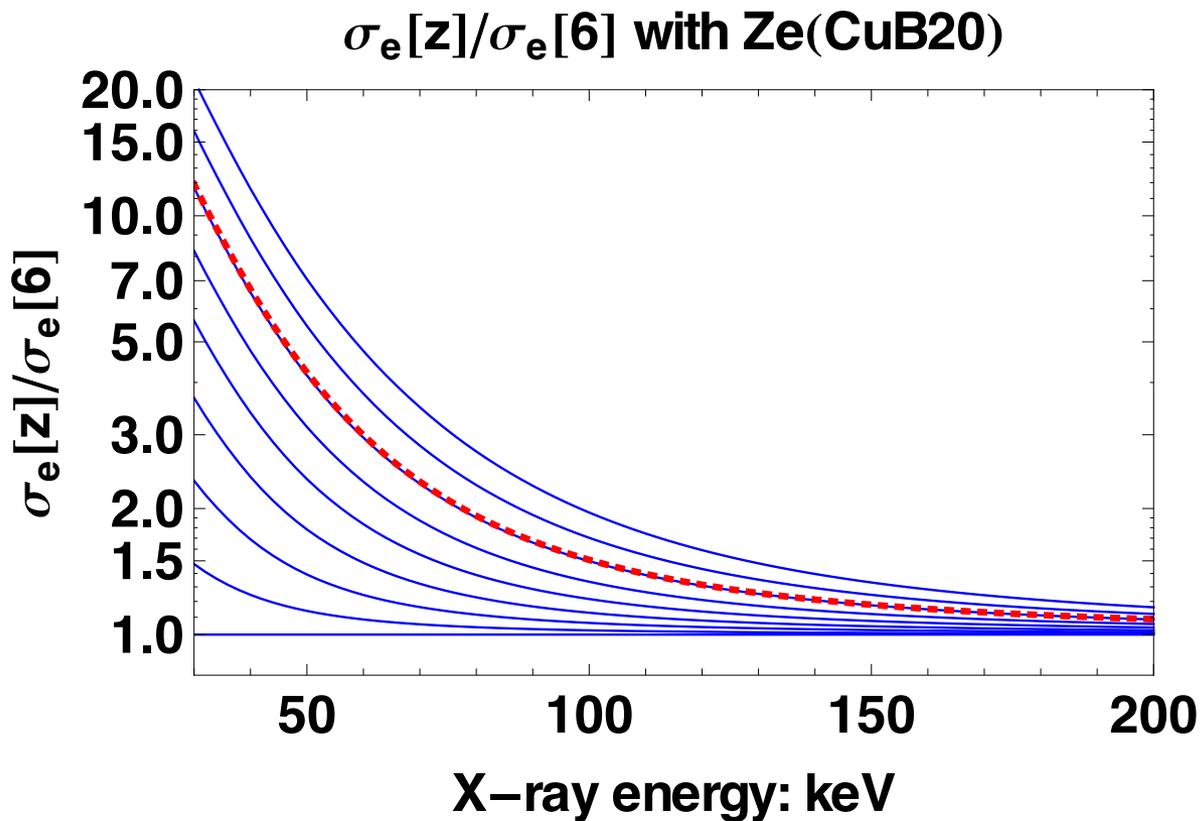
**Transmission Error CuB₂₀ and KCl
Cross-Section Fit vs. Transmission Fit**



- Transmission is calculated through 2.5 g/cm² of material



- To optimize the Ze fit over a range of energies, a flat spectrum, $S[Ex]$, was used.
- To test the sensitivity of the resulting transmission fit to spectrum, two “typical” spectra (100 kV and 160 kV) were used.



- The red Z_e for CuB_{20} interpolates the cross-section just slightly above the blue $Z = 18$ curve.

Use of a “Transmission Fit Criterion Minimizes Transmission Errors

			Net Transmission Error (%) by Spectrum		
Material	Ze	Fit Criterion	Flat	100 kV	160 kV
CuB ₂₀	17.85	$\sum (\sigma_{e-mat} - \sigma_{Ze})^2$ => minimum	0.566	0.556	0.690
CuB ₂₀	18.12	$\sum (\text{Trans}_{e-mat} - \text{Trans}_{Ze})^2$ => minimum	0.007	- 0.121	0.096
KCl	18.11	$\sum (\text{Trans}_{e-mat} - \text{Trans}_{Ze})^2$ => minimum	0.004	0.003	0.006

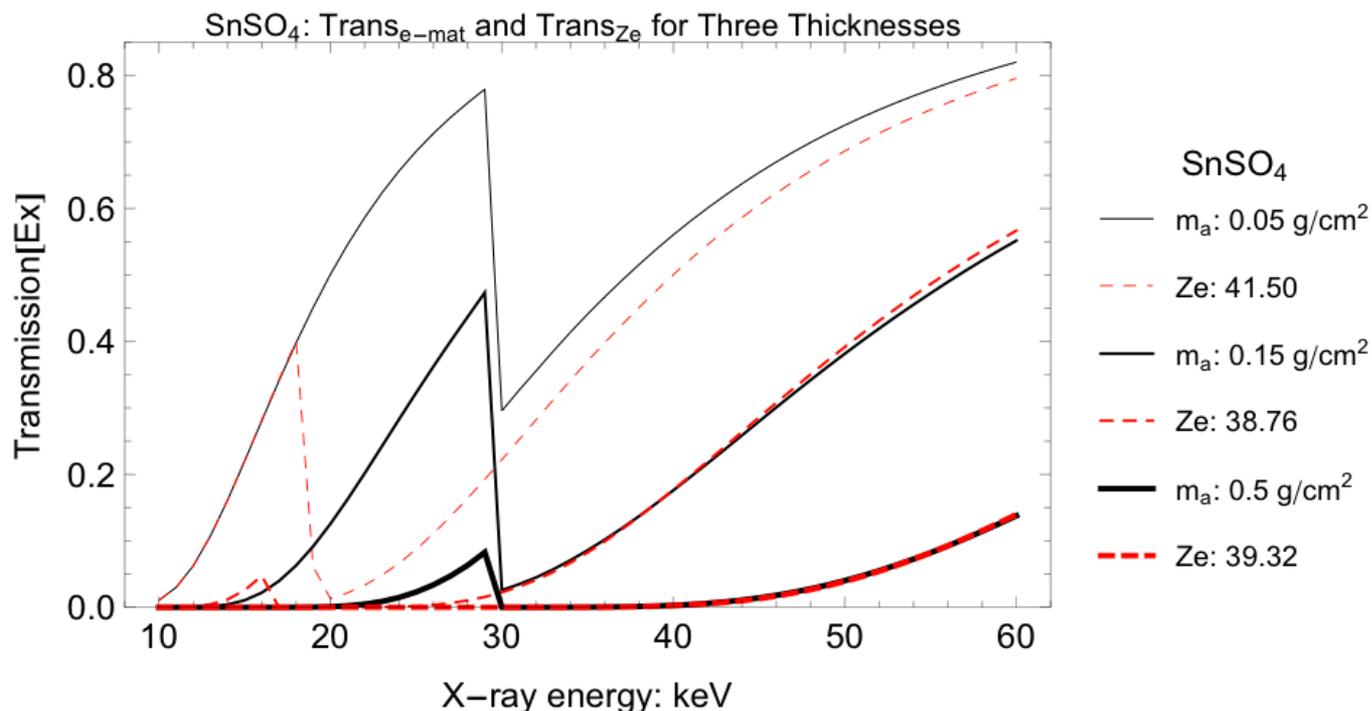
- **Table 1.** Using a “best fit to cross-section” finds a Ze of 17.85 for CuB₂₀, and greater than 1/2% errors in calculating the transmission through a layer that is 2.5 g/cm² in areal density. Transmission errors are summed over the spectrum, S[Ex].
- In the second row, the transmission error is greatly reduced by using a “best fit to transmission criterion.
- For KCL, both criteria provided the same Ze, and and a minimal transmission error.
- For reference, the calculated transmissions for the Flat, 100 kV, and 160 kV spectra are 42.7%, 17.0 %, and 54.8 % respectively.

- The fit of Z_e to a particular application is adjusted by selecting a spectral range, $S[Ex]$, and a representative-layer areal density (m_m or m_e).
- This adjustment, provides the best approximation for the context of the particular application, but is only effective if the Z_e values are relatively insensitive to the exact values of m_m (or m_e) and $S[Ex]$.
- Table 2 demonstrates that Z_e is indeed insensitive to those values over a broad selection of spectral range and m_m .

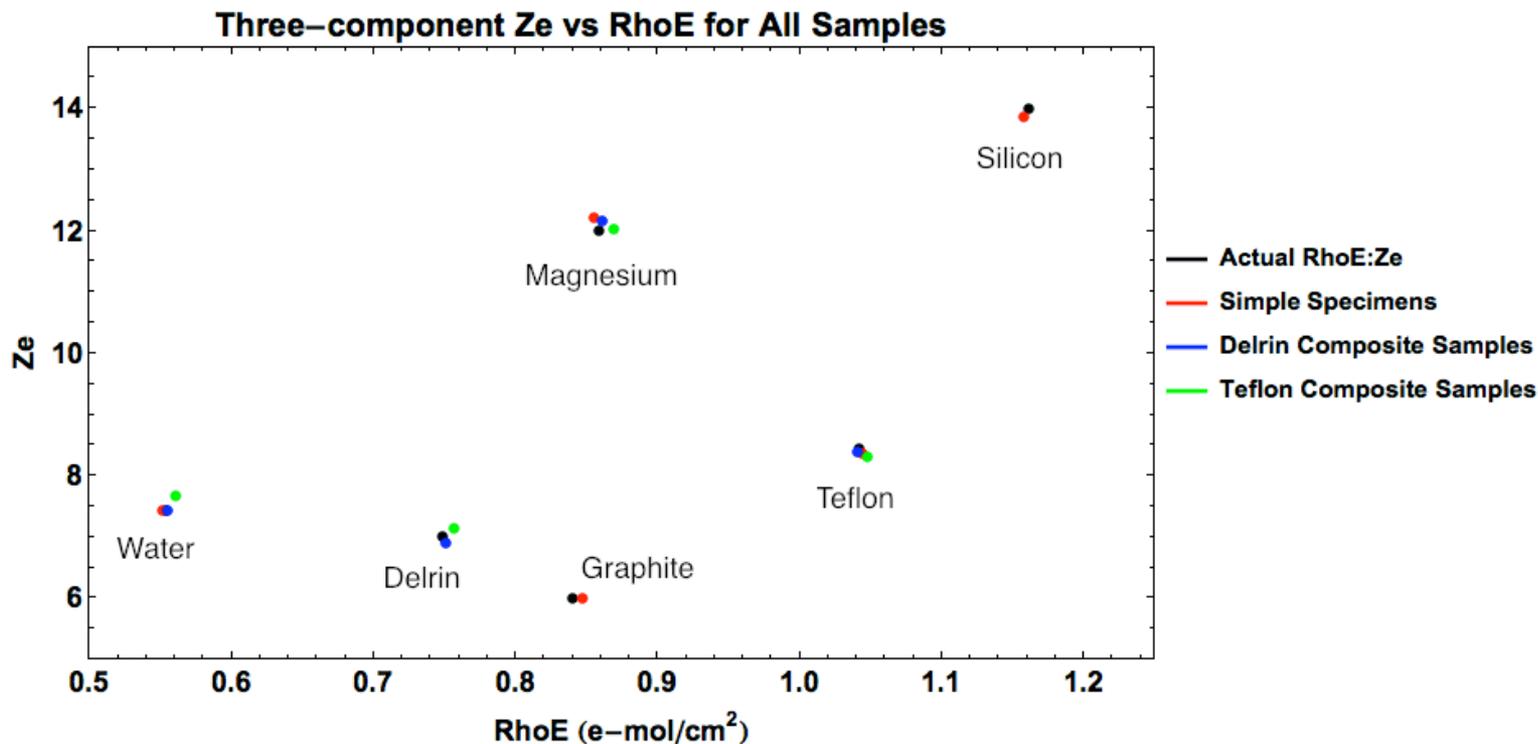


		Areal Density (g/cm ²)		
Energy Range (keV)	Material	0.5	2.5	12.5
10 – 60	CH ₄	5.14	5.12	5.07
	SiO ₂	11.63	11.64	11.64
	SnSO ₄	N/A ^b	N/A ^b	N/A ^b
30 – 160	CH ₄	5.05	5.05	5.03
	SiO ₂	11.64	11.64	11.61
	SnSO ₄	39.80	40.13	40.24
80 - 450	CH ₄	4.83	4.83	4.82
	SiO ₂	11.58	11.58	11.57
	SnSO ₄	39.99	40.25	40.46

- **Table 2.** Ze was calculated for three materials, three different “representative-layer” thicknesses and three spectral regions. The Ze were relatively insensitive to these parameters.
- ^b SnSO₄ was not an appropriate candidate for Ze at these energies (next slide)



- X-ray absorption edges can cause a discontinuity in the absorption/transmission of spectra through a material. This behavior cannot in general be approximated except by the specific element causing the discontinuity



- An example of Ze and ρ_e (RhoE) used to discriminate materials is illustrated by recent data showing a three-spectrum decomposition of a variety of simple and complex specimens.

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ZeCalc Algorithm Details

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- ZeCalc (a Java app.) calculates Ze based on the transmission-fit criterion for:
 - $10 \text{ keV} \leq E_x \leq 500 \text{ keV}$, including 3 tungsten bremsstrahlung spectra
 - $1 \leq Z_e \leq 100$
 - $0.5 \leq \text{mm} \leq 12.5 \text{ g/cm}^2$
 - Outputs: Ze, optional ρ_e , Graphs of $Z[E_x]$, Transmission vs. E_x , Transmission Error vs. E_x .
 - For more information, contact the authors.

- Ze provides a practical tool for obtaining an accurate approximation to the energy-dependent cross section for transmission measurements using polychromatic spectra.
- This tool can be used for quantitative measurements on compounds and mixtures over energies from 10–1000 keV.
- A more detailed technical discussion of this material will be submitted for publication in *Nuclear Instruments and Methods Part B*. in the near future.

1. W.V. Mayneord, Acta of the International Union against Cancer ii (1937) 275.
2. S. Gowda, S. Krishnaveni, T. Yashoda, T.K. Umesh, R. Gowda, Pramana, J. Phys. 6 (3) (2004) 529.
3. S.V. Naydenov, V.D. Ryzhikov, C.F. Smith, Nucl. Instr. and Meth. B 215 (2004) 552.
4. S.R. Manohara, S.M. Hanagodimath, K.S. Thind, L. Gerward, Nucl. Instr. and Meth. B 266 (2008) 3906.
5. M.L. Taylor, R.L. Smith, F. Dossing, R.D. Franich, Med. Phys. 39 (4) (2012) 1769.
6. D.E. Cullen, J.H. Hubbell, L. Kissel, EPDL97: The Evaluated Photon Data Library '97 Version UCRL-LR-50400 6 (Rev. 5) (1997).
7. J.H. Hubbell, S.M. Seltzer, <http://www.nist.gov/pml/data/xraycoef/index.cfm>, (2009), accessed Jan. 31, 2013.
8. B.L. Henke, E.M. Gullickson, and J.C. Davis, Atomic Data And Nuclear Data Tables 54 no.2, 181–342 (1993).